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**Phet Gas  
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# **Simulation Answer Key**

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## Simulation

Pump gas molecules to a box and see what happens as you change the volume, add or remove heat, and more. Measure the temperature and pressure, and discover how the properties of the gas vary in relation to each other. Examine kinetic energy and speed histograms for light and heavy particles. Explore diffusion and determine how

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concentration,  
temperature, mass,  
and radius affect the  
rate of ...

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Simulation.

advertisement ... Be  
sure to include the  
variables and  
constants in your  
answer. Part 4. Gay-  
Lussac's Law Purpose:  
to determine the  
relationship between

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pressure and temperature (with everything else constant). Procedure: You get to design how to test the purpose. Write down the settings you put the simulator to.

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Worksheet Answers

## **Phet Gas Properties**

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## Answer Key

### **Gas Laws Simulation Activity Answer Key**

This simulation is intended for students to practice completing the gas law calculations, as well as to connect specific particle behavior with the associated variable of volume, temperature or pressure. In addition, by clicking on the “Add Data” button, students

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## Simulation

can collect data to  
create a graph for each  
of the gas laws.

### **Classroom Resources | Simulation Activity: Gas Laws | AACT**

Be sure to include the variables and constants in your answer. (your answer should contain the words temperature, volume, pressure and # of particles.) Part 3. Charles' Law, Purpose:

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to determine the  
relationship between  
volume and  
temperature (with  
everything else  
constant). ... PHet Gas  
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(Regnault)  $\rightarrow p =$   
constant Variables  $T \rightarrow$

$V$  Gay-Lussac  $\rightarrow V =$   
constant Variables  $T \rightarrow$

$p$  Boyle (-Mariotte)  $\rightarrow T$   
 $=$  constant  $V \rightarrow p$

Avogadro  $\rightarrow p =$   
constant  $T =$  constant  
Variables  $N \rightarrow V.$

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Kylie Murry: HS MS:  
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Kinetic Theory of Gases  
Lecture Demo: Bob  
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## Simulation **PDF Free Download**

Without using the simulation, sketch what you think the graphs would look like.

Note: Be sure to label your x and y axes. Assume that other variables remain constant (for example, temperature and number of particles remains constant in the first graph).

## **Solved: Gas Law Physics Lab**

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